

ORGANIC SYNTHESIS WITH REAGENTS DERIVED FROM $3R_3SiMgMe$ AND $MnCl_2$

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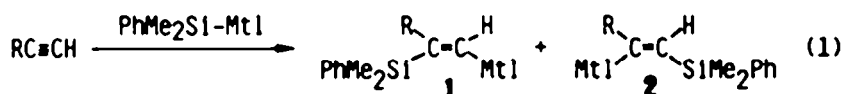
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(Received in UK 29 September 1987)

Summary: Synthetically useful reactions mediated by reagents derived from 3 equiv. of $R_3SiMgMe$ and $MnCl_2$ are disclosed. (1) The manganese species reacted with terminal acetylenes to give 1,2-disilylated 1-alkenes. Mono- and bis(trimethylsilyl)acetylenes gave tri- and tetrasilylated ethenes, respectively, in good yields. Highly strained tetrakis(trimethylsilyl)ethene has now become easily accessible by this technique. (2) The reaction of alkenyl halides, alkenyl sulfides, and enol phosphates with the title reagents provides vinylsilanes in good yields. The method is also applicable to the synthesis of allylsilanes from allylic sulfides and ethers. (3) Treatment of 1,3-dienes with the manganese reagents provides silylated allylmanganese compounds which add to the carbonyl moiety with high regioselectivity.

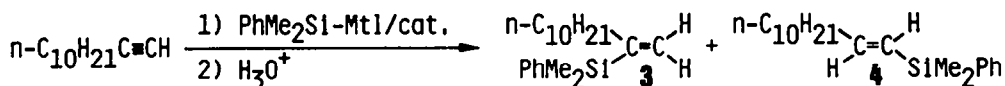
Introduction

In connection with an investigation of the stereoselective synthesis of the side chain of a plant growth steroidal hormone, brassinolide, we required a method for the formation of alkenylmetal species (1) containing trialkylsilyl group. Regio- and stereoselective addition of Si-Metal reagent² to acetylenes was an attractive possibility to obtain this compound (eq. 1). Although exclusive formation of 2-metallo-1-silyl-1-alkenes (2) by silylaluminum,³ silylcupration⁴ or silyltitanation⁵ was previously reported, the selective generation of 1-metallo-2-silyl isomers 1 hitherto has not been described to our knowledge.



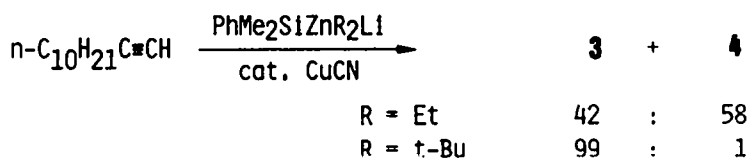
First we investigated simultaneous addition of a silyl group and a metal to acetylenes with regioselectivity using $PhMe_2SiLi$ and metal compounds such as

MeMgI, Et₂AlCl, ZnBr₂, and Et₃B, in the presence of transition-metal catalysts.^{6,7} Platinum- or copper-catalyzed silylmagnesiation (PhMe₂SiMgMe) followed by aqueous quenching provided exclusively (E)-1-silyl-1-alkenes, which have the same regiochemistry as the products prepared from stoichiometric silylcupration or silyltitanation. On the other hand, palladium catalyzed silyl-alumination (PhMe₂SiAlEt₂) afforded the regioisomer 2-silyl-1-alkenes with high regioselectivity.⁸



MeMgI-cis-PtCl ₂ (P- <i>n</i> -Bu ₃) ₂ or CuI	<1	:	>99
Et ₂ AlCl- PdCl ₂ (P(<i>o</i> -CH ₃ C ₆ H ₄) ₃) ₂	85	:	15

Dialkyl(dimethylphenylsilyl)zinc-lithium⁹ added effectively to internal acetylenes as well as to terminal ones to provide vinylsilanes with high stereo- and regioselectivities. The regioselectivity of the reaction was strongly affected by the nature of the dialkylzinc employed. The use of a bulky alkyl group on zinc favours the formation of 2-silyl-1-alkenes.



Organocuprates have been extensively studied and widely applied to organic synthesis.¹⁰ In contrast, little attention has been paid to the related reactions of triorganomanganese compounds, although the addition of 3 equiv of BuLi to MnCl₂ is known to give Bu₃MnLi which adds to α,β-unsaturated carbonyl compounds in 1,4-fashion.¹¹ Here we describe the behaviour of silylmanganese reagents which are derived from 3 equiv of R₃SiMgMe and MnCl₂ and are believed to be (R₃Si)₃MnMgMe. The reactions involve (1) acetylenes, (2) alkenyl halides, alkenyl sulfides, and enol phosphates, and (3) dienes as the acceptor components. The formulation of the reagents, for example as (R₃Si)₃MnMgMe, has no evidence to support it and represents the stoichiometry of the components rather than an explicit structural formula.

Results and Discussion

(1) Disilylation of Acetylenes with Si-Mn Reagent

Palladium(0) catalyzed reaction of acetylenes with disilanes such as HMe₂SiSiMe₂H, FMe₂SiSiMe₂F, and (MeO)Me₂SiSiMe₂(OMe) produces the corresponding disilylated olefins in good yield.¹² In contrast, hexamethyldisilane gives less than 10% yield of doubly silylated products.

We have examined the reaction of the Si-Mn reagent with acetylenes¹³ and observed the rather unexpected formation of disilylated products.¹⁴ (Trimethylsilyl)lithium¹⁵ was treated with methylmagnesium iodide and anhydrous manganese(II) chloride.¹⁶ Addition of 1-alkyne to the resulting reagent provided an E, Z mixture of 1,2-bis(trimethylsilyl)-1-alkene. The reaction proceeded smoothly with silylacetylenes as well as terminal acetylenes to give

tri- and tetrasilylated ethenes.

Although the role of $MeMgI$ is not clear, its presence is essential for the formation of disilylated products. Without $MeMgI$, monosilylated products were obtained predominantly after aqueous workup. For instance, treatment of 4-(benzyloxy)-1-butyne with $3Me_3SiLi-MnCl_2$ gave a mixture of 4-(benzyloxy)-2-(trimethylsilyl)-1-butene, 4-(benzyloxy)-1-(trimethylsilyl)-1-butene, and disilylated product in 1:1:1 ratio (65% combined yield). The reaction has been extended to distannylation of acetylenes.¹⁷ The representative results are shown in Table 1.

Treatment of 5-(benzyloxy)-2-pentyne with the reagent which is believed to be $(Me_3Si)_3MnMgMe$ (**5a**) gave 2,3-bis(trimethylsilyl)-2-alkene **6a** in 78% yield. Meanwhile, the addition of H_2O (or D_2O , MeI) to the reaction mixture after stirring at $0^\circ C$ for 20 min without warming up to room temperature gave monosilylated product **6b** (71%) (or **6c** (71%), **6d** (70%)) along with the disilylated one **6a** (13-20%).¹⁸ Thus, the formation of disilylated alkenes may be explained as follows: (1) Addition of the reagent **5a** to the triple bond in a *cis* fashion giving silylated alkenylmanganese **A** and (2) reductive elimination of manganese affording disilylated olefin. The *cis* addition of the Si-Mn component was confirmed by the following experiment. A monosilylated alkene, 5-(benzyloxy)-2-(dimethylphenylsilyl)-2-pentene was prepared from $(PhMe_2Si)_3MnMgMe$ (**5b**) and 5-

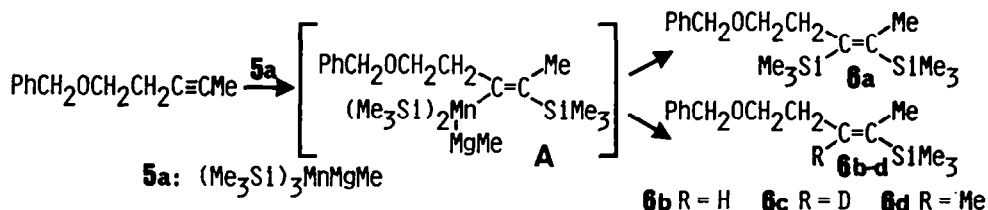
Table 1. Disilylation or distannylation of acetylenes with Si-Mn or Sn-Mn reagent

$$RC\equiv CR^1 \xrightarrow[\text{(Me}_3\text{SnLi, MeMgI, MnCl}_2\text{)}]{\text{R}^2_3\text{SiLi, MeMgI, MnCl}_2} \begin{matrix} R & & R^1 \\ & \diagdown & / \\ & C=C & \\ & / & \diagdown \\ R^2_3Si & & SiR^2_3 \\ \text{(Me}_3\text{Sn)} & & \text{(SnMe}_3\text{)} \end{matrix}$$

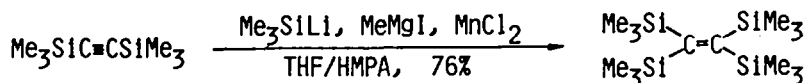
Acetylene $RC\equiv CR^1$	Reagent R^2_3Si	Yield(%)	Product Z/E
$HC\equiv CSiMe_3$	Me_3SiLi	72	—
$n-C_6H_{13}C\equiv CH$	Me_3SiLi	66	33/67
$PhC\equiv CH$	Me_3SiLi	65	50/50
$n-C_6H_{13}C\equiv CSiMe_3$	Me_3SiLi	80	—
$PhCH_2OCH_2CH_2C\equiv CH$	Me_3SiLi	55	50/50
$PhCH_2OCH_2CH_2C\equiv CH$	$PhMe_2SiLi$	51	42/58
$PhCH_2OCH_2CH_2C\equiv CD$	$PhMe_2SiLi$	55	42/58 ^{a)}
$PhCH_2OCH_2CH_2C\equiv CSiMe_3$	Me_3SiLi	59	—
$THPOCH_2CH_2C\equiv CH$	Me_3SiLi	63	65/35
$THPOCH_2CH_2C\equiv CH$	$PhMe_2SiLi$	70	65/35
$THPOCH_2CH_2C\equiv CSiMe_3$	Me_3SiLi	83	—
$HOCH_2CH_2C\equiv CSiMe_3$	Me_3SiLi ^{b)}	58	—
$n-C_{10}H_{21}C\equiv CH$	Me_3SnLi ^{c)}	56	100/0
$n-C_{10}H_{21}C\equiv CD$	Me_3SnLi	55	100/0
$THPOCH_2CH_2C\equiv CH$	Me_3SnLi	54	100/0
$THPOCH_2CH_2C\equiv CH$	$n-Bu_3SnLi$ ^{c)}	48	90/10
$PhCH_2OCH_2CH_2C\equiv CH$	Me_3SnLi	52	90/10

a) Deuterium was retained completely, thus reagents did not cause the acetylenic proton-metal exchange. b) Three millimoles of manganese reagent and 1.0 mmol of substrate were employed. c) Prepared from $SnCl_2$ and 3 equiv of alkylolithium. See Ref. 20.

(benzyloxy)-2-pentyne in the same way as **6b**. Protodesilylation with $n\text{-Bu}_4\text{NF}^{18}$ gave 5-(benzyloxy)-2-pentene. The ^1H NMR spectra proved that the olefin had \underline{Z} configuration (>95%, $\underline{J} = 11$ Hz). Similar desilylation of **6b** failed.¹⁹



The reaction can be successfully applied to the synthesis of highly strained tetrakis(trimethylsilyl)ethene, which is not readily available by known methods.²⁰



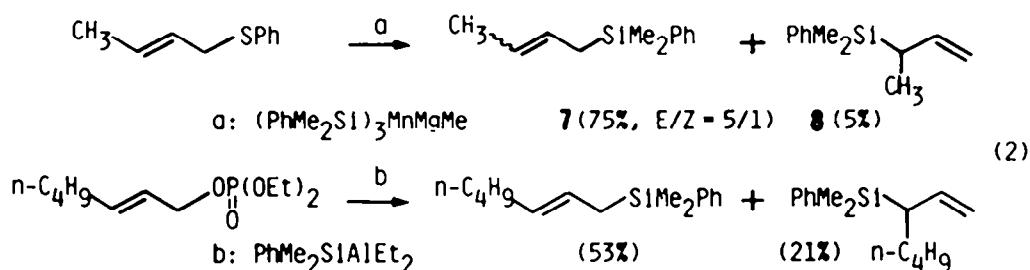
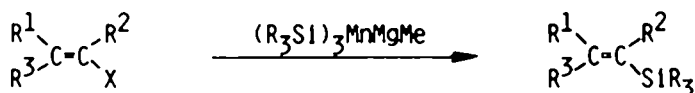
(2) Synthesis of Vinylsilanes and Allylsilanes by Cross-coupling of Manganese Reagents with Alkenyl and Allylic Compounds


The organoaluminium reagent, $\text{PhMe}_2\text{SiAlEt}_2$, is effective for the transformation of enol phosphates or allylic phosphates into vinylsilanes or allylsilanes,²¹ respectively.²² Treatment of alkenyl halides (or alkenyl sulfides, or enol phosphates) or allylic sulfides (or allylic ethers)^{23,24} with the Si-Mn reagent provides vinylsilanes or allylsilanes in an alternative way. Representative results are summarized in Table 2.

Points to be noted are as follows. (1) The reaction proceeded stereospecifically. Whereas (\underline{E})-1-iodo-1-dodecene gave (\underline{E})-dimethylphenylsilyl-1-dodecene exclusively upon treatment with the reagent **5b**, (\underline{Z})-1-iodo-1-dodecene provided the (\underline{Z}) vinylsilane preferentially. In the latter case, low reaction temperature such as -95°C was essential in order to obtain high stereospecificity (Entry 4). (2) Not only alkenyl halides²⁵ but also alkenyl sulfides and enol phosphates were easily converted into the corresponding vinylsilanes. A vinyl ether such as 1-methoxycyclohexene reacted sluggishly with the manganese reagent to provide a trace of the desired vinylsilane (<5%). (3) The use of MeMgI was essential for the reaction. The reaction of cyclohexenyl diethyl phosphate with $(\text{R}_3\text{Si})_3\text{MnLi}$ gave cyclohexanone, possibly due to the nucleophilic attack of the reagent on the phosphorus atom. (4) The amount of Li_2MnCl_4 could be reduced to a catalytic amount. For instance, treatment of diethylcyclohexenyl phosphate with $\text{PhMe}_2\text{SiMgMe}$ in the presence of 10 mol% of Li_2MnCl_4 gave the desired vinylsilane in 66% yield.

Allylic sulfides also smoothly coupled with **5b** to give the corresponding allylic silanes in good yields.^{26,27} The results are shown in Table 3. The reaction proceeded with high regioselectivity. Both (\underline{E})-1-phenylthio-2-butene and 3-phenylthio-1-butene gave a 95:5 mixture of 1-dimethylphenylsilyl-2-butene (**7**) and the regioisomer, 3-dimethylphenylsilyl-1-butene (**8**). The silylanion appears to attack the less hindered site of the allylic system. Geranyl benzyl sulfide afforded geranylsilane exclusively. The selectivity is much higher than the reaction of allylic phosphates with $\text{PhMe}_2\text{SiAlEt}_2$ ²² (eq. 2). Allylic ethers and halides such as 1-benzyloxy-2-butene, 1-tert-butyldimethylsiloxy-2-butene, and crotyl bromide easily reacted with the manganese reagent to give the corre-

corresponding allylsilanes.


 Table 2. Synthesis of vinylsilanes with Si-Mn reagents^a)


Entry	R ¹	Substrate			Reaction Conditions		Product	
		R ²	R ³	X	Temp (°C)	Time (h)	Y (%) ^b	E/Z
(PhMe₂Si)₃MnMgMe								
1	Me(H)	H	H(Me)	Br ^c	0	3	100	47/53
2	n-C ₆ H ₁₃	H	H	I	0	0.5	88	100/0
3	n-C ₁₀ H ₂₁	H	H	I	0	0.5	72	100/0
4	H	H	n-C ₁₀ H ₂₁	I	0	0.5	73	20/80
5	H	H	n-C ₁₀ H ₂₁	I	-95	1.3	40 ^d	0/100
6	H	H	Ph	Br	25	1	70	39/61
7	H	H	Ph	Br	-78	3	50	0/100
8	Me	H	Me	Br	0	0.5	70	—
9	n-C ₁₀ H ₂₁	H	Me	I	0	2	75 ^e	100/0
10	n-C ₅ H ₁₁	n-C ₅ H ₁₁	Me	I	0	2	47 ^f	80/20
11	-(CH ₂) ₄ -	H	H	Br	0	2	93	—
12					0	2	65	—
13	Ph	H	H	SMe	0	1	70	100/0
14	-(CH ₂) ₄ -	H	H	SMe	0	1	75	—
15	H	n-C ₆ H ₁₃	H	OP(O)(OPh) ₂	25	5	74	—
16	-(CH ₂) ₄ -	H	H	OP(O)(OEt) ₂	0	1	66	—
17	H	n-C ₃ H ₇	H	OSO ₂ CF ₃	0	1	99	—
(Me₃Si)₃MnMgMe								
18	H	n-C ₁₀ H ₂₁	H	I	0	3	100	—
19	n-C ₁₀ H ₂₁	H	H	I	-95	2	100	30/70
20	-(CH ₂) ₆ -	H	H	SPh	0	3	89	—

a) Manganese reagent (1.5 mmol) and substrate (1.0 mmol) were employed. b) Isolated yields. c) Commercially available 1-bromo-1-propene (Aldrich Chemical Co.) was used (E/Z = 40/60). d) Starting material was recovered (30%). e) 2-Methyl-1-dodecene was obtained as byproduct (5% yield). f) (Z)-6-Methyl-6-dodecene was also obtained (25% yield).

Table 3. Synthesis of allylsilanes by cross coupling of allylic sulfides and ethers with Si-Mn reagent^{a)}

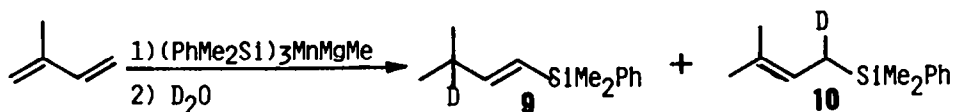
Starting Material	Product Yield(%) ^{b)}	Starting Material	Product Yield(%) ^{b)}
	 (91)		 (84) ^{c, d)}
	 (60)		 (98)
	7 (88) ^{e)} 8 (2)		
	 (89)		
	 (56) ^{c)}	$R'' = \text{CH}_2\text{Ph}$	7 (65) ^{f)} 8 (5)
		$R'' = \text{SiMe}_2^t\text{Bu}$	7 (71) ^{g)} 8 (2)
			7 (80) 8 (2)

a) Reactions were performed at 0°C for 30 min. Two mmol of the reagent and one mmol of substrate were employed. b) Isolated yields. c) $(\text{Me}_3\text{Si})_3\text{MnMgMe}$ was used instead of $(\text{PhMe}_2\text{Si})_3\text{MnMgMe}$. d) Reaction mixture was stirred for 15 h. e) $E/Z = 2/1$. f) $E/Z = 20/1$. g) $E/Z = 20/1$.

(3) Silylmanganation of 1,3-dienes

In the first section of this paper, we have reported addition reactions of $(\text{R}_3\text{Si})_3\text{MnMgMe}$ reagents to acetylenic compounds to give disilylated alkenes. In further extension, we have examined the reaction of 1,3-dienes with $(\text{R}_3\text{Si})_3\text{MnMgMe}$ and observed the selective formation of vinylsilanes after quenching with carbonyl compounds.^{28,29}

Treatment of isoprene with the reagent $(\text{PhMe}_2\text{Si})_3\text{MnMgMe}$ gave a mixture of (*E*)-3-deuterio-1-dimethylphenylsilyl-3-methyl-1-butene (**9**) and 1-deuterio-1-dimethylphenylsilyl-3-methyl-2-butene (**10**) (**9**:**10** = 1:5) after quenching with D_2O . Table 4 summarizes further examples. Quenching with MeI gave a mixture of (*E*)-3,3-dimethyl-1-dimethylphenylsilyl-1-butene **11** ($E = \text{Me}$) and 4-dimethylphenylsilyl-2-methyl-2-pentene **12** ($E = \text{Me}$) (**11**:**12** = 1:2) in 90% combined yield. In contrast, addition of aldehyde or ketone to the silylated allylmanganese intermediate gave (*E*)-vinylsilane **11** as a single product.



The reagent **5b** possibly adds to isoprene in 1,2-fashion (or 1,4 fashion) to give the hypothetical intermediates **B** and **C**, the latter being produced by β -

elimination of Mn-H and successive readdition. Alicyclic dienes such as cyclopentadiene and 1,3-cyclooctadiene were reluctant to react. The reagents $PhMe_2SiLi$ or $PhMe_2SiMgMe$ did not react with isoprene and the manganese atom played a key role in the reaction. The addition of $MeMgI$ was essential for the complete isomerization of **B** to **C**. Treatment of isoprene with $(PhMe_2Si)_3MnLi$ derived from $PhMe_2SiLi$ and $MnCl_2$ gave 4-dimethylphenylsilyl-2-methyl-1-butene (30%) arising from **B** along with **11** (E = H, 10%) and **12** (E = H, 39%) after quenching with water.

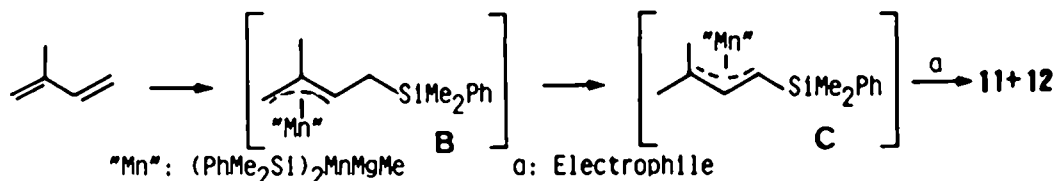
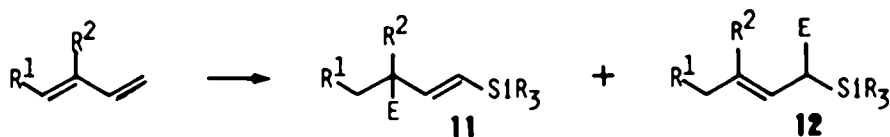


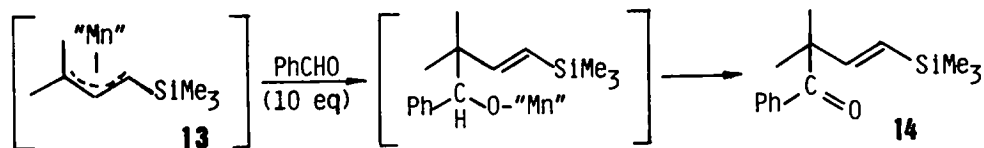
Table 4. Silylmanganation of 1,3-dienes with Si-Mn reagent



Entry	Diene		Reagent	Electrophile	Product		Y(%)	
	R ¹	R ²			11	12		
1	H	H	$(PhMe_2Si)_3MnMgMe$	D_2O	44	25		
2				MeI	11	69		
3				PhCHO	70 ^{a)}	0		
4				PhCH=CHCHO	44 ^{b)}	0		
5				CH ₃ CH=CHCHO	88 ^{b)}	0		
6				Me ₂ CO	95	0		
7				$(Me_3Si)_3MnMgMe$	PhCHO	82	0	
8					CH ₃ CH=CHCHO	50 ^{b)}	0	
9	H	Me	$(PhMe_2Si)_3MnMgMe$	D_2O	13	65		
10				MeI	30	60		
11				n-BuCHO	85	0		
12				PhCHO	78	0		
13				PhCH=CHCHO	74 ^{b)}	0		
14				$(Me_3Si)_3MnMgMe$	n-C ₈ H ₁₇ CHO	52	0	
15					PhCHO	77	0	
16			CH ₃ CH=CHCHO		75 ^{b)}	0		
17			Me ₂ CO		76	0		
18	Me	H	$(PhMe_2Si)_3MnMgMe$	D_2O	46	46		
19				PhCHO	68	0		

a) Erythro/threo = 1/1. b) Only 1,2-adducts were obtained.

The presence of a large excess (10 equiv) of the carbonyl electrophiles such as PhCHO and PhCH=CHCHO caused Oppenauer type oxidation to give ketones as the main products. For instance, treatment of a solution of **13** (1.0 mmol) derived from isoprene and **5a** with PhCHO (10 mmol) afforded 2,2-dimethyl-1-phenyl-4-(trimethylsilyl)-3-buten-1-one (**14**) in 68% yield along with 2,2-dimethyl-1-phenyl-4-(trimethylsilyl)-3-buten-1-ol (24%) and PhCH₂OH.



Experimental

Distillation of the products were performed by use of Kugelrohr (Büchi), and boiling points are indicated by an air bath temperature without correction. ¹H-NMR spectra were taken on a Varian XL-200 spectrometer, CDCl₃ was used as solvent, chemical shifts being given in δ with tetramethylsilane as an internal standard. IR spectra were determined on a JASCO IR-810 spectrometer. The analyses were carried out by the staff at the Elemental Analyses Center of Kyoto University. Tetrahydrofuran was freshly distilled from sodium benzophenone ketyl.

General Procedure for the Generation of Silylmanganate Reagents. An ethereal solution of MeMgI (1.1 M, 4.1 ml, 4.5 mmol) was added to a THF solution of Me₂PhSiLi³⁰ (0.75 M, 6.0 ml, 4.5 mmol) (or a THF-HMPA (10 ml, 4:1) solution of Me₃SiLi (prepared in situ from 6.0 mmol of hexamethyldisilane and 4.5 mmol of MeLi-LiBr complex¹⁵) at 0 °C under an argon atmosphere. After stirring for 15 min, a THF solution of Li₂MnCl₄³¹ (1.0 M, 1.5 ml, 1.5 mmol) was added. The resulting dark brown mixture was stirred for an additional 15 min before use.

General Procedure for the Disilylation of Acetylenes. A solution of tetrahydropyranyl ether of 4-(trimethylsilyl)-3-buten-1-ol (0.23 g, 1.0 mmol) in THF (3.0 ml) was added to a solution of (Me₃Si)₃MnMgMe (1.5 mmol) in THF-HMPA and the resulting mixture was stirred for 3 h at 0 °C. The mixture was diluted with ether and poured into saturated NH₄Cl. Purification by preparative tlc on silica-gel gave tetrahydropyranyl ether of 3,4,4-tris(trimethylsilyl)-3-buten-1-ol (0.31 g) in 83% yield as a colourless oil: bp 150 °C (1.0 Torr, bath temp); IR (neat) 1440, 1350, 1260, 1250, 1200, 1140, 1120, 1030, 840 cm⁻¹; NMR (CDCl₃) δ 0.19 (s, 9H), 0.22 (s, 9H), 0.23 (s, 9H), 1.5-1.9 (m, 1H), 2.89 (t, *J* = 8 Hz, 2H), 3.36 (dt, *J* = 8, 10 Hz, 1H), 4.61 (t, *J* = 3 Hz, 1H); MS, *m/z* 372 (M⁺, 1), 270 (64), 197 (43), 155 (48), 85 (100), 73 (99). Found: C, 58.02; H, 11.06%. Calcd for C₁₈H₄₀O₂Si₃: C, 58.00; H, 10.82%.

Tris(trimethylsilyl)ethene: bp 85 °C (15 Torr, bath temp); IR (neat) 2950, 1250, 900, 840, 680 cm⁻¹; NMR (CDCl₃) δ 0.09 (s, 9H), 0.16 (s, 9H), 0.17 (s, 9H), 7.34 (s, 1H); MS, *m/z* 244 (M⁺, 3), 229 (2), 156 (11), 73 (100), 45 (14). Found: C, 54.20; H, 11.79%. Calcd for C₁₁H₂₈Si₃: C, 54.02; H, 11.54%.

(E)-1,2-Bis(trimethylsilyl)-1-octene: bp 90 °C (2 Torr, bath temp); IR (neat) 2950, 2930, 2850, 1250, 840, 760, 690 cm⁻¹; NMR (CDCl₃) δ 0.07 (s, 9H), 0.11 (s, 9H), 0.89 (t, *J* = 4 Hz, 3H), 1.25-1.37 (m, 8H), 2.21-2.27 (m, 2H), 5.97 (s, 1H); MS, *m/z* 256 (M⁺, 1), 241 (6), 186 (40), 167 (23), 98 (39), 73 (100). Analytical sample was prepared by preparative glc (*t_r* = 3 min, Silicone OV 17, 3% on Uniport HP, 2 m, 120 °C). Found: C, 65.53; H, 12.76%. Calcd for C₁₄H₃₂Si₂: C, 65.54; H, 12.57%.

(Z)-1,2-Bis(trimethylsilyl)-1-octene: bp 90 °C (2 Torr, bath temp); IR (neat)

2950, 2930, 2850, 1250, 840, 760, 690 cm^{-1} ; NMR ($CDCl_3$) δ 0.13 (s, 9H), 0.15 (s, 9H), 0.88 (t, $J = 4$ Hz, 3H), 1.4-1.6 (m, 8H), 2.17 (dt, $J = 1, 4$ Hz, 2H), 6.27 (d, $J = 1$ Hz, 1H); MS, m/z 241 ($M^+ - CH_3$, 6), 186 (20), 167 (16), 98 (35), 73 (100); $t_R = 5$ min, Silicone OV 17, 3% on Uniport HP 2m, 120 °C. Found: C, 65.65; H, 12.82%. Calcd for $C_{14}H_{32}Si_2$: C, 65.54; H, 12.57%.

1,1,2-Tris(trimethylsilyl)-1-octene: bp 120 °C (3 Torr, bath temp); IR (neat) 2950, 2920, 1470, 1410, 1380, 1250, 870, 840 cm^{-1} ; NMR ($CDCl_3$) δ 0.19 (s, 18H), 0.20 (s, 9H), 0.89 (t, $J = 7$ Hz, 3H), 1.21-1.36 (m, 8H), 2.50 (t, $J = 9$ Hz, 2H); MS, m/z 328 (M^+ , 8), 313 (8), 255 (41), 167 (100), 155 (34), 131 (37), 73 (51). Found: C, 62.29; H, 12.37%. Calcd for $C_{17}H_{40}Si_3$: C, 62.11; H, 12.26%.

(E)-1,2-Bis(trimethylsilyl)-4-benzyloxy-1-butene: bp 120 °C (1 Torr, bath temp); IR (neat) 2950, 2890, 2850, 1450, 1360, 1250, 1100, 840 cm^{-1} ; NMR ($CDCl_3$) δ 0.05 (s, 9H), 0.11 (s, 9H), 2.63 (t, $J = 8$ Hz, 2H), 3.41 (t, $J = 8$ Hz, 2H), 4.52 (s, 2H), 6.11 (s, 1H), 7.26-7.37 (m, 5H); MS, m/z 291 ($M^+ - CH_3$, 1), 200 (4), 179 (17), 147 (5), 112 (32), 91 (100), 73 (71). Pure sample was obtained by preparative glc, $t_R = 6$ min, Silicone OV 17, 3% on Uniport HP 2 m, 170 °C. Found: C, 66.60; H, 9.98%. Calcd for $C_{17}H_{30}Si_2$: C, 66.60; H, 9.86%.

(Z)-1,2-Bis(trimethylsilyl)-4-benzyloxy-1-butene: bp 120 °C (1 Torr, bath temp); IR (neat) 2950, 2900, 2850, 1450, 1360, 1250, 1100, 860, 830, 750, 730, 690 cm^{-1} ; NMR ($CDCl_3$) δ 0.13 (s, 9H), 0.15 (s, 9H), 2.52 (dt, $J = 1, 8$ Hz, 2H), 3.47 (t, $J = 8$ Hz, 2H), 4.51 (s, 2H), 6.38 (t, $J = 1$ Hz, 1H), 7.27-7.36 (m, 5H); MS, m/z 291 ($M^+ - CH_3$, 1), 179 (20), 147 (6), 112 (33), 91 (100), 73 (71); $t_R = 10$ min, Silicone OV 17, 3% on Uniport HP 2 m, 170 °C. Found: C, 66.85; H, 10.10%. Calcd for $C_{17}H_{30}Si_2$: C, 66.60; H, 9.86%.

1,2-Bis(dimethylphenylsilyl)-4-benzyloxy-1-butene (E/Z = 1/1): bp 160 °C (0.3 Torr, bath temp); IR (neat) 2950, 1430, 1250, 1110, 840, 820, 730, 700 cm^{-1} ; NMR ($CDCl_3$) δ 0.19 (s, 3H), 0.25 (s, 3H), 0.35 (s, 3H), 0.39 (s, 3H), 2.58 (t, $J = 8$ Hz, 1H), 2.60 (t, $J = 8$ Hz, 1H), 4.18 (s, 1H), 4.44 (s, 1H), 6.40 (s, 0.5H), 6.72 (s, 0.5H), 7.2-7.6 (m, 15H); MS, m/z 415 ($M^+ - CH_3$, 0.2), 324 (5), 135 (100), 91 (60). Found: C, 75.38; H, 8.11%. Calcd for $C_{27}H_{34}OSi_2$: C, 75.29; H, 7.96%.

4-Benzyloxy-1,1,2-tris(trimethylsilyl)-1-butene: bp 140 °C (1 Torr, bath temp); IR (neat) 1450, 1360, 1260, 1250, 1100, 870, 830 cm^{-1} ; NMR ($CDCl_3$) δ 0.16 (s, 9H), 0.17 (s, 9H), 0.19 (s, 9H), 2.90 (t, $J = 7$ Hz, 2H), 3.38 (t, $J = 7$ Hz, 2H), 4.49 (s, 2H), 7.3-7.4 (m, 5H); MS, m/z 278 (M^+ , 2), 205 (4), 199 (9), 155 (21), 143 (19), 91 (100), 73 (73). Found: C, 63.48; H, 10.33%. Calcd for $C_{20}H_{38}Si_3$: C, 63.42; H, 10.11%.

Tetrahydropyranyl Ether of (E)-3,4-Bis(trimethylsilyl)-3-buten-1-ol: bp 105 °C (3 Torr, bath temp); IR (neat) 2950, 1250, 1140, 1120, 1040, 840 cm^{-1} ; NMR ($CDCl_3$) δ 0.08 (s, 9H), 0.14 (s, 9H), 1.5-1.6 (m, 6H), 2.62 (t, $J = 8$ Hz, 2H), 3.34 (dt, $J = 8, 8$ Hz, 1H), 3.4-3.55 (m, 1H), 3.69 (td, $J = 8, 10$ Hz, 1H), 3.8-3.95 (m, 2H), 4.61 (t, $J = 3$ Hz, 1H), 6.12 (s, 1H); MS, m/z 227 ($M^+ - SiMe_3$, 1), 173 (11), 85 (100), 73 (41); $t_R = 4$ min, Silicone OV 17 3% on Uniport HP 2 m, 160 °C. Found: C, 59.70; H, 10.96%. Calcd for $C_{15}H_{32}O_2Si_2$: C, 59.94; H, 10.73%.

Tetrahydropyranyl Ether of (Z)-3,4-Bis(trimethylsilyl)-3-buten-1-ol: bp 105 °C (3 Torr, bath temp); IR (neat) 2900, 1250, 1140, 1120, 1040, 860, 840 cm^{-1} ; NMR ($CDCl_3$) δ 0.13 (s, 9H), 0.17 (s, 9H), 1.5-1.6 (m, 6H), 2.49 (dt, $J = 1, 8$ Hz, 2H), 3.38 (dt, $J = 8, 8$ Hz, 1H), 3.4-3.55 (m, 1H), 3.66 (td, $J = 8, 10$ Hz, 1H), 3.8-3.95 (m, 2H), 4.60 (t, $J = 3$ Hz, 1H), 6.40 (t, $J = 1$ Hz, 1H); MS, m/z 285 ($M^+ - CH_3$, 0.1), 201 (7), 173 (22), 85 (100), 73 (42); $t_R = 6.5$ min, Silicone OV 17 3% on Uniport HP 2 m, 160 °C. Found: C, 60.10; H, 10.96%. Calcd for $C_{15}H_{32}O_2Si_2$: C, 59.94; H, 10.73%.

Tetrahydropyranyl Ether of 3,4-Bis(dimethylphenylsilyl)-3-buten-1-ol (E/Z = 2/1): bp 160 °C (0.3 Torr, bath temp); IR (neat) 2950, 1430, 1250, 1110, 1030, 840, 820, 700 cm^{-1} ; NMR (CDCl_3) E-isomer: δ 0.39 (s, 6H), 0.42 (s, 6H), 1.4-1.6 (m, 6H), 3.09 (dt, $J = 8, 10$ Hz, 1H), 3.3-3.6 (m, 3H), 3.6-3.9 (m, 2H), 4.24 (t, $J = 3$ Hz, 1H), 6.41 (s, 1H), 7.2-7.4 (m, 3H), 7.4-7.6 (m, 2H). Z-isomer: δ 0.18 (s, 6H), 0.28 (s, 6H), 1.4-1.6 (m, 6H), 3.1 (m, 1H), 3.3-3.6 (m, 3H), 3.6-3.9 (m, 2H), 4.54 (t, $J = 3$ Hz, 1H), 6.73 (s, 1H), 7.2-7.4 (m, 3H), 7.4-7.6 (m, 2H); MS, m/z 409 ($\text{M}^+ - \text{CH}_3$, 0.1), 235 (50), 135 (64), 85 (100). Found: C, 70.92; H, 8.77%. Calcd for $\text{C}_{25}\text{H}_{36}\text{O}_2\text{Si}_2$: C, 70.70; H, 8.54%.

(Z)-Tetrahydropyranyl Ether of 1,2-Bis(trimethylstannyl)-3-buten-1-ol: bp 110 °C (0.1 Torr, bath temp); IR (neat) 2950, 2920, 1350, 1120, 1030, 730 cm^{-1} ; NMR (CDCl_3) δ 0.16 (s, 9H), 0.19 (s, 9H), 1.5-1.8 (m, 6H), 3.62 (dt, $J = 1, 5$ Hz, 2H), 3.4-3.5 (m, 2H), 3.7-3.9 (m, 2H), 4.59 (t, $J = 3$ Hz, 1H), 6.75 (d, $J = 1$ Hz, 1H).

2,3-Bis(dimethylphenylsilyl)-5-benzyloxy-2-pentene: bp 120 °C (1 Torr, bath temp); IR (neat) 2920, 1240, 1090, 840, 830 cm^{-1} ; NMR (CDCl_3) δ 0.15 (s, 9H), 0.16 (s, 9H), 1.85 (s, 3H), 2.67 (t, $J = 7$ Hz, 2H), 3.34 (t, $J = 7$ Hz, 2H), 4.51 (s, 2H), 7.25-7.35 (m, 5H); MS, m/z 305 ($\text{M}^+ - \text{CH}_3$, 0.1), 179 (53), 91 (99), 73 (100). Found: C, 67.39; H, 10.33%. Calcd for $\text{C}_{18}\text{H}_{32}\text{O}_2\text{Si}_2$: C, 67.43; H, 10.06%.

(E)-5-Benzyloxy-2-trimethylsilyl-2-pentene: bp 100 °C (2 Torr, bath temp); IR (neat) 2950, 2850, 1620, 1250, 1100, 840 cm^{-1} ; NMR (CDCl_3) δ 0.04 (s, 9H), 1.68 (dt, $J = 1, 1$ Hz, 3H), 2.44 (qdt, $J = 1, 6, 7$ Hz, 2H), 3.51 (t, $J = 7$ Hz, 2H), 4.54 (s, 2H), 5.73 (qt, $J = 1, 6$ Hz, 1H), 7.27-7.37 (m, 5H); MS, m/z 233 ($\text{M}^+ - \text{CH}_3$, 1), 143 (20), 99 (20), 91 (100), 73 (42). Found: C, 72.52; H, 10.02%. Calcd for $\text{C}_{15}\text{H}_{24}\text{OSi}$: C, 72.52; H, 9.74%.

(E)-5-Benzyloxy-3-deuterio-2-trimethylsilyl-2-pentene: bp 100 °C (2 Torr, bath temp); IR (neat) 2950, 2850, 1600, 1250, 1100, 840 cm^{-1} ; NMR (CDCl_3) δ 0.04 (s, 9H), 1.68 (s, 3H), 2.43 (t, $J = 7$ Hz, 2H), 3.51 (t, $J = 7$ Hz, 2H), 4.53 (s, 2H), 7.27-7.37 (m, 5H); MS, m/z 234 ($\text{M}^+ - \text{CH}_3$, 1), 144 (11), 100 (12), 91 (100), 73 (61). Found: C, 72.21; H, 10.31%. Calcd for $\text{C}_{15}\text{H}_{23}\text{DOSi}$: C, 72.23; H, 10.10%.

(E)-5-Benzyloxy-3-methyl-2-trimethylsilyl-2-pentene: bp 160 °C (1 Torr, bath temp); IR (neat) 2900, 2800, 1600, 1440, 1350, 1240, 1090, 830 cm^{-1} ; NMR (CDCl_3) δ 0.11 (s, 9H), 1.5 (m, 3H), 1.8 (m, 3H), 2.46 (t, $J = 9$ Hz, 2H), 3.50 (t, $J = 9$ Hz, 2H), 4.52 (s, 2H), 7.3 (m, 5H); MS, m/z 247 ($\text{M}^+ - \text{CH}_3$, 1), 179 (37), 156 (25), 91 (65), 73 (100). Found: C, 73.09; H, 10.07%. Calcd for $\text{C}_{16}\text{H}_{26}\text{OSi}$: C, 73.22; H, 9.98%.

Tetrakis(trimethylsilyl)ethene. The compound was obtained following the general procedure described above (76% yield). The physical data of the sample were identical with those in the literature.²⁰ mp 102 °C; $^1\text{H-NMR}$ (CDCl_3) δ 0.06 (s, 36H); $^{13}\text{C-NMR}$ (CDCl_3) δ 4.3, 195.4; MS, m/z 316 (M^+ , 1), 155 (56), 131 (23), 73 (100). Found: m/z 316.1873. Calcd for $\text{C}_{14}\text{H}_{36}\text{Si}_4$: M^+ , 316.1895.

The Reaction between Silylmanganate Reagents and Hetero Substituted Olefinic Compounds. Following procedure is representative. A THF solution of 1-cyclohexenyl diethyl phosphate (0.12 g, 0.50 mmol) was added to a THF solution of $(\text{PhMe}_2\text{Si})_3\text{MnMgMe}$ (1.5 mmol) and resulting mixture was stirred for 2 h at room temperature. Workup and purification by preparative tlc gave 1-(dimethylphenylsilyl)-1-cyclohexene (70 mg, 65% yield): bp 67 °C (2 Torr, bath temp); IR (neat) 2924, 1615, 1428, 1247, 1112, 1062, 827, 807, 769, 728, 698, 650 cm^{-1} ; NMR (CDCl_3) δ 0.31 (s, 6H), 1.5-1.7 (m, 4H), 1.9-2.1 (m, 4H), 6.0-6.1 (m, 1H), 7.3-7.4 (m, 3H), 7.5-7.6 (m, 2H). Found: C, 77.81; H, 9.47%. Calcd for $\text{C}_{14}\text{H}_{20}\text{Si}$: C, 77.71; H, 9.32%. The physical data of 1-(dimethylphenylsilyl)-1-propene,³² (E) and (Z)-2-(dimethylphenylsilyl)styrene^{4,5}, and 1-(dimethylphenyl-

silyl)-2-methyl-1-dodecene¹⁹ were identical with those of authentic samples.

6-(Dimethylphenylsilyl)-7-methyl-6-dodecene (E:Z = 80:20): bp 133 °C (0.1 Torr, bath temp); IR (neat) 2954, 2930, 2854, 1466, 1428, 1247, 1110, 830, 814, 770, 728, 698 cm^{-1} ; NMR (CDCl_3) δ 0.24 (s, 1.2H), 0.34 (s, 4.8H), 0.7-1.0 (m, 6H), 1.0-1.5 (m, 12H), 1.61 (s, 2.4H), 1.72 (s, 0.6H), 1.9-2.2 (m, 4H), 7.3-7.4 (m, 3H), 7.4-7.6 (m, 2H). Found: C, 79.62; H, 11.60%. Calcd for $\text{C}_{21}\text{H}_{36}\text{Si}$: C, 79.67; H, 11.46%.

(E)-1-(Dimethylphenylsilyl)-1-octene: bp 66 °C (2 Torr, bath temp); IR (neat) 2954, 2924, 2852, 1617, 1428, 1247, 1114, 989, 840, 821, 728, 698 cm^{-1} ; NMR (CDCl_3) δ 0.31 (s, 6H), 0.88 (t, $J = 6.8$ Hz, 3H), 1.2-1.5 (m, 8H), 2.13 (ddt, $J = 6.2, 1.4, 6.0$ Hz, 2H), 5.76 (dt, $J = 18.5, 1.4$ Hz, 1H), 6.12 (dt, $J = 18.5, 6.2$ Hz, 1H), 7.3-7.4 (m, 3H), 7.5-7.6 (m, 2H). Found: C, 77.74; H, 10.77%. Calcd for $\text{C}_{16}\text{H}_{26}\text{Si}$: C, 77.97; H, 10.63%.

2-(Dimethylphenylsilyl)-1-octene: bp 80 °C (2 Torr, bath temp); IR (neat) 2954, 2924, 2852, 1459, 1428, 1248, 1111, 924, 832, 816, 774, 729, 698 cm^{-1} ; NMR (CDCl_3) δ 0.36 (s, 6H), 0.85 (t, $J = 6.8$ Hz, 3H), 1.1-1.7 (m, 8H), 2.11 (tm, $J = 7.6$ Hz, 2H), 5.39 (dt, $J = 3.0, 1.0$ Hz, 1H), 5.67 (dt, $J = 3.0, 1.5$ Hz, 1H), 7.3-7.4 (m, 3H), 7.5-7.6 (m, 2H). Found: C, 77.75; H, 10.90%. Calcd for $\text{C}_{16}\text{H}_{26}\text{Si}$: C, 77.97; H, 10.63%.

2-(Dimethylphenylsilyl)-1-pentene: bp 70 °C (2 Torr, bath temp); IR (neat) 2954, 2924, 2854, 1460, 1428, 1248, 1111, 831, 728, 697 cm^{-1} ; NMR (CDCl_3) δ 0.36 (s, 6H), 0.83 (t, $J = 7.1$ Hz, 3H), 1.2-1.4 (m, 4H), 2.11 (tm, $J = 7.5$ Hz, 2H), 5.39 (dt, $J = 3.0, 1.0$ Hz, 1H), 5.68 (dt, $J = 3.0, 1.5$ Hz, 1H), 7.3-7.4 (m, 3H), 7.5-7.6 (m, 2H). Found: C, 76.12; H, 10.07%. Calcd for $\text{C}_{13}\text{H}_{20}\text{Si}$: C, 76.40; H, 9.86%.

2-(Trimethylsilyl)-1-dodecene: bp 95 °C (2 Torr, bath temp); IR (neat) 2952, 2922, 2852, 1466, 1459, 1247, 922, 836, 754, 688 cm^{-1} ; NMR (CDCl_3) δ 0.08 (s, 9H), 0.88 (t, $J = 6.8$ Hz, 3H), 1.27 (br.s, 14H), 1.2-1.5 (m, 2H), 2.12 (t, $J = 6.3$ Hz, 2H), 5.30 (dt, $J = 3.1, 1.0$ Hz, 1H), 5.55 (dt, $J = 3.1, 1.5$ Hz, 1H). Found: C, 74.90; H, 13.69%. Calcd for $\text{C}_{15}\text{H}_{32}\text{Si}$: C, 74.91; H, 13.41%.

5-(Dimethylphenylsilyl)-4-nonene: bp 123 °C (2 Torr, bath temp); IR (neat) 2954, 2924, 2856, 1611, 1466, 1459, 1427, 1247, 1109, 830, 814, 770, 728, 698 cm^{-1} ; NMR (CDCl_3) δ 0.33 (s, 6H), 0.82 (t, $J = 6.6$ Hz, 3H), 0.90 (t, $J = 6.8$ Hz, 3H), 1.19 (br.s, 6H), 1.3-1.4 (m, 4H), 2.0-2.2 (m, 4H), 5.80 (t, $J = 6.9$ Hz, 1H), 7.3-7.4 (m, 3H), 7.5-7.6 (m, 2H). Found: C, 78.86; H, 11.44%. Calcd for $\text{C}_{19}\text{H}_{32}\text{Si}$: C, 79.09; H, 11.18%.

The Reaction of Silylmanganate Reagents with Conjugated Dienes. A typical procedure is as follows. To a THF solution of $(\text{PhMe}_2\text{Si})_3\text{MnMgMe}$ (1.5 mmol) was added a THF solution of 1,3-butadiene (1.5 M, 1.0 ml, 1.5 mmol) at 0 °C and the mixture was stirred for 15 min. Benzaldehyde (0.16 g, 1.5 mmol) was added to the solution and stirring was continued for 1 h at room temperature. The reaction mixture was diluted by hexane and 0.5 ml of saturated aqueous ammonium chloride was added. After additional stirring for 10 min, the supernatant was decanted, dried over anhydrous sodium sulfate, and the solvent was removed in vacuo. Purification by preparative tlc on silica-gel afforded (E)-4-(dimethylphenylsilyl)-2-methyl-1-phenyl-3-buten-1-ol (1:1 mixture of diastereoisomers): bp 140 °C (2 Torr, bath temp); IR (neat) 3370 (broad), 2956, 1612, 1453, 1427, 1248, 1112, 1019, 998, 842, 730, 698 cm^{-1} ; NMR (CDCl_3) δ 0.26 (s, 1.5H), 0.27 (s, 1.5H), 0.34 (s, 3.0H), 0.88 (d, $J = 6.8$ Hz, 1.5H), 1.04 (d, $J = 6.8$ Hz, 1.5H), 2.01 (br.s, 0.5H), 2.12 (br.s, 0.5H), 2.53 (ddq, $J = 7.6, 7.1, 6.8$ Hz, 0.5H), 2.61 (ddq, $J = 6.6, 5.8, 6.8$ Hz, 0.5H), 4.40 (d, $J = 7.6$ Hz, 0.5H), 4.56 (d, $J = 5.8$ Hz, 0.5H), 5.74 (dd, $J = 18.9, 1.1$ Hz, 0.5H), 5.95 (d, $J = 18.8$ Hz, 0.5H), 5.99 (dd, $J = 18.9, 6.6$ Hz, 0.5H), 6.10 (dd, $J = 18.8, 7.1$ Hz, 0.5H),

7.2-7.6 (m, 10H). Found: C, 77.10; H, 8.36%. Calcd for $C_{19}H_{24}OSi$: C, 76.97; H, 8.16%.

A 1:6 Mixture of (E)-1-(Dimethylphenylsilyl)-3-methyl-1-butene and 4-(Dimethylphenylsilyl)-2-pentene: bp 115 °C (22 Torr, bath temp); IR (neat) 2956, 1428, 1249, 1113, 834, 776, 729, 699 cm^{-1} ; NMR ($CDCl_3$) δ 0.25 (s, 2.58H), 0.27 (s, 2.58H), 0.32 (s, 0.84H), 1.00 (d, $J = 7.0$ Hz, 3.42H), 1.46 (dd, $J = 6.5, 1.4$ Hz, 2.58H), 2.09 (dq, $J = 10.5, 7.2$ Hz, 0.86H), 2.2-2.4 (m, 0.14H), 5.1-5.5 (m, 1.72H), 5.71 (dd, $J = 18.7, 1.4$ Hz, 0.14H), 6.10 (dd, $J = 18.7, 5.8$ Hz, 0.14H), 7.3-7.4 (m, 3H), 7.5-7.6 (m, 2H). Found: C, 76.59; H, 9.97%. Calcd for $C_{13}H_{20}Si$: C, 76.40; H, 9.86%.

(1E, 5E)-6-(Dimethylphenylsilyl)-4-methyl-1-phenyl-1,5-hexadien-3-ol (1:1 mixture of diastereoisomers): bp 160 °C (2 Torr, bath temp); IR (neat) 3348 (br), 2958, 1613, 1450, 1428, 1249, 1113, 966, 843, 828, 787, 762, 731, 693, 663 cm^{-1} ; NMR ($CDCl_3$) δ 0.32 (s, 1.5H), 0.33 (s, 1.5H), 0.35 (s, 3H), 1.07 (d, $J = 6.8$ Hz, 1.5H), 1.10 (d, $J = 6.8$ Hz, 1.5H), 2.4-2.6 (m, 1H), 4.10 (dd, $J = 6.9, 6.3$ Hz, 0.5H), 4.2-4.3 (m, 0.5H), 5.91 (dd, $J = 18.7, 0.9$ Hz, 0.5H), 5.96 (d, $J = 18.8$ Hz, 0.5H), 6.0-6.3 (m, 2H), 6.57 (dd, $J = 16.0, 1.1$ Hz, 0.5H), 7.2-7.5 (m, 8H), 7.5-7.7 (m, 2H). Found: C, 78.47; H, 8.24%. Calcd for $C_{21}H_{26}OSi$: C, 78.21; H, 8.13%.

(1E, 5E)-1-(Dimethylphenylsilyl)-3-methyl-1,5-heptadien-4-ol (1:1 mixture of diastereoisomers): bp 88 °C (2 Torr, bath temp); IR (neat) 3352 (br), 2956, 1613, 1428, 1248, 1114, 998, 966, 842, 824, 730, 698 cm^{-1} ; NMR ($CDCl_3$) δ 0.32 (s, 3H), 0.33 (s, 3H), 0.99 (d, $J = 7.4$ Hz, 1.5H), 1.04 (d, $J = 6.4$ Hz, 1.5H), 1.6-1.7 (m, 3H), 2.2-2.5 (m, 1H), 3.82 (dd, $J = 7.8, 7.2$ Hz, 0.5H), 3.96 (dd, $J = 6.8, 6.0$ Hz, 0.5H), 5.4-5.8 (m, 2H), 5.85 (dd, $J = 19.0, 1.1$ Hz, 0.5H), 5.92 (d, $J = 19.0$ Hz, 0.5H), 6.05 (dd, $J = 19.0, 6.9$ Hz, 0.5H), 6.08 (dd, $J = 19.0, 6.6$ Hz, 0.5H), 7.3-7.4 (m, 3H), 7.4-7.6 (m, 2H). Found: C, 73.82; H, 9.54%. Calcd for $C_{16}H_{24}OSi$: C, 73.79; H, 9.29%.

(E)-2,3-Dimethyl-5-(dimethylphenylsilyl)-4-penten-2-ol: bp 90 °C (2 Torr, bath temp); IR (neat) 3372 (br), 2968, 1612, 1428, 1371, 1248, 1114, 843, 825, 728, 698 cm^{-1} ; NMR ($CDCl_3$) δ 0.34 (s, 6H), 1.05 (d, $J = 6.9$ Hz, 3H), 1.17 (d, $J = 4.3$ Hz, 3H), 1.56 (br.s, 1H), 2.27 (ddq, $J = 7.7, 0.6, 6.9$ Hz, 1H), 5.88 (dd, $J = 18.7, 0.6$ Hz, 1H), 6.10 (dd, $J = 18.7, 7.7$ Hz, 1H), 7.3-7.4 (m, 3H), 7.5-7.6 (m, 2H). Found: C, 72.79; H, 9.95%. Calcd for $C_{15}H_{24}OSi$: C, 72.52; H, 9.74%.

(E)-2-Methyl-1-phenyl-4-(trimethylsilyl)-3-buten-1-ol (60:40 diastereomeric mixture): bp 91 °C (2 Torr, bath temp); IR (neat) 3368 (br), 2954, 1612, 1453, 1247, 1020, 996, 865, 837, 699 cm^{-1} ; NMR ($CDCl_3$) δ 0.01 (s, 5.4H), 0.07 (s, 3.6H), 0.85 (d, $J = 6.8$ Hz, 1.2H), 0.99 (d, $J = 6.8$ Hz, 1.8H), 1.97 (br.s, 0.6H), 2.14 (br.s, 0.4H), 2.48 (ddq, $J = 7.9, 7.1, 6.8$ Hz, 0.4H), 2.54 (dddq, $J = 6.5, 5.5, 1.2, 6.8$ Hz, 0.6H), 4.36 (d, $J = 7.9$ Hz, 0.4H), 4.61 (d, $J = 5.5$ Hz, 0.6H), 5.65 (dd, $J = 18.8, 1.2$ Hz, 0.6H), 5.84 (d, $J = 18.8$ Hz, 0.4H), 5.92 (dd, $J = 18.8, 6.5$ Hz, 0.6H), 5.97 (dd, $J = 18.8, 7.1$ Hz, 0.4H), 7.2-7.4 (m, 5H). Found: C, 71.66; H, 9.68%. Calcd for $C_{14}H_{22}OSi$: C, 71.73; H, 9.46%.

(1E, 5E)-3-Methyl-1-(trimethylsilyl)-1,5-heptadien-4-ol (57:43 diastereomeric mixture): bp 65 °C (2 Torr, bath temp); IR (neat) 3362 (br), 2954, 1613, 1248, 995, 965, 866, 837 cm^{-1} ; NMR ($CDCl_3$) δ 0.06 (s, 5.13H), 0.06 (s, 3.87H), 0.97 (d, $J = 6.8$ Hz, 1.29H), 1.01 (d, $J = 6.9$ Hz, 1.71H), 1.71 (dm, $J = 7.2$ Hz, 3H), 2.22 (ddq, $J = 7.2, 7.0, 6.8$ Hz, 0.43H), 2.34 (ddq, $J = 6.8, 6.0, 6.9$ Hz, 0.57H), 3.79 (dd, $J = 7.4, 7.2$ Hz, 0.43H), 3.95 (dd, $J = 6.1, 6.0$ Hz, 0.57H), 5.4-5.5 (m, 1H), 5.6-5.8 (m, 1H), 5.73 (dd, $J = 18.8, 1.1$ Hz, 0.57H), 5.79 (d, $J = 18.8$ Hz, 0.43H), 5.92 (dd, $J = 18.8, 7.0$ Hz, 0.43H), 5.97 (dd, $J = 18.8, 6.8$ Hz, 0.57H). Found: C, 66.36; H, 11.48%. Calcd for $C_{11}H_{22}OSi$: C, 66.60; H, 11.18%.

A 1:2 Mixture of (E)-3,3-dimethyl-1-(dimethylphenylsilyl)-1-butene and 2-Methyl-4-(dimethylphenylsilyl)-2-pentene: bp 65 °C (2 Torr, bath temp); IR (neat) 2956, 2922, 2856, 1427, 1247, 1113, 830, 729, 697 cm^{-1} ; NMR ($CDCl_3$) δ 0.22 (s, 2.01H), 0.25 (s, 2.01H), 0.31 (s, 1.98H), 0.97 (d, $J = 7.2$ Hz, 2.01H), 1.02 (s, 2.97H), 1.44 (d, $J = 1.2$ Hz, 2.01H), 1.68 (d, $J = 1.0$ Hz, 2.01H), 1.93 (dq, $J = 10.9, 7.3$ Hz, 0.67H), 4.93 (dm, $J = 10.9$ Hz, 0.67H), 5.66 (d, $J = 19.0$ Hz, 0.33H), 6.13 (d, $J = 19.0$ Hz, 0.33H), 7.3-7.5 (m, 3H), 7.5-7.6 (m, 2H). Found: C, 76.89; H, 10.41%. Calcd for $C_{14}H_{22}Si$: C 76.99; H, 10.15%.

(E)-3,3-Dimethyl-1-(dimethylphenylsilyl)-1-octen-4-ol: bp 100 °C (2 Torr, bath temp); IR (neat) 3450 (br), 2954, 2930, 2860, 1466, 1248, 1113, 999, 844, 827, 731, 698 cm^{-1} ; NMR ($CDCl_3$) δ 0.34 (s, 6H), 0.91 (t, $J = 7.1$ Hz, 3H), 1.02 (s, 6H), 1.2-1.8 (m, 7H), 3.28 (br.d, $J = 9.0$ Hz, 1H), 5.83 (d, $J = 19.5$ Hz, 1H), 6.08 (d, $J = 19.5$ Hz, 1H), 7.3-7.5 (m, 3H), 7.5-7.7 (m, 2H). Found: C, 74.17; H, 10.60%. Calcd for $C_{18}H_{30}OSi$: C, 74.42; H, 10.41%.

(E)-2,2-Dimethyl-4-(dimethylphenylsilyl)-1-phenyl-3-buten-1-ol: bp 140 °C (2 Torr, bath temp); IR (neat) 3420 (br), 2958, 1611, 1453, 1427, 1248, 1113, 999, 844, 826, 731, 699 cm^{-1} ; NMR ($CDCl_3$) δ 0.34 (s, 6H), 0.98 (s, 3H), 1.03 (s, 3H), 1.93 (br.s, 1H), 4.45 (s, 1H), 5.81 (d, $J = 19.1$ Hz, 1H), 6.17 (d, $J = 19.1$ Hz, 1H), 7.26 (br.s, 5H), 7.3-7.5 (m, 3H), 7.5-7.6 (m, 2H). Found: C, 77.21; H, 8.59%. Calcd for $C_{20}H_{26}OSi$: C, 77.36; H, 8.44%.

(1E, 5E)-4,4-Dimethyl-6-(dimethylphenylsilyl)-1-phenyl-1,5-hexadien-3-ol: bp 150 °C (2 Torr, bath temp); IR (neat) 3440 (br), 2958, 2868, 1670, 1612, 1494, 1450, 1427, 1248, 1113, 998, 968, 844, 824, 735, 697 cm^{-1} ; NMR ($CDCl_3$) δ 0.35 (s, 6H), 1.08 (s, 6H), 1.68 (br.s, 1H), 4.01 (d, $J = 6.6$ Hz, 1H), 5.88 (d, $J = 19.1$ Hz, 1H), 6.19 (d, $J = 19.1$ Hz, 1H), 6.21 (dd, $J = 16.5, 6.9$ Hz, 1H), 6.91 (dd, $J = 16.5, 1.0$ Hz, 1H), 7.2-7.4 (m, 8H), 7.5-7.6 (m, 2H). Found: C, 78.36; H, 8.53%. Calcd for $C_{22}H_{28}OSi$: C, 78.51; H, 8.39%.

(E)-3,3-Dimethyl-1-(trimethylsilyl)-1-dodecen-4-ol: bp 124 °C (2 Torr, bath temp); IR (neat) 3450 (br), 2954, 1247, 1113, 999, 844, 826, 730, 698 cm^{-1} ; NMR ($CDCl_3$) δ 0.06 (s, 9H), 0.88 (t, $J = 6.3$ Hz, 3H), 0.98 (s, 6H), 1.28 (br.s, 12H), 1.3-1.5 (m, 2H), 1.55 (br.s, 1H), 5.70 (d, $J = 19.2$ Hz, 1H), 5.97 (d, $J = 19.2$ Hz, 1H). Found: C, 71.51; H, 12.84%. Calcd for $C_{17}H_{36}OSi$: C, 71.76; H, 12.75%.

(E)-2,2-Dimethyl-1-phenyl-4-(trimethylsilyl)-3-buten-1-ol: bp 113 °C (2 Torr, bath temp); IR (neat) 3448 (br), 2952, 1609, 1453, 1248, 1041, 1026, 1001, 866, 837, 731, 700 cm^{-1} ; NMR ($CDCl_3$) δ 0.07 (s, 9H), 0.95 (s, 3H), 1.00 (s, 3H), 5.70 (d, $J = 19.2$ Hz, 1H), 6.06 (d, $J = 19.2$ Hz, 1H), 7.2-7.4 (m, 5H). Found: C, 72.27; H, 9.94%. Calcd for $C_{15}H_{24}OSi$: C, 72.52; H, 9.74%.

(E)-3,3-Dimethyl-1-(trimethylsilyl)-1,5-heptadien-4-ol: bp 87 °C (2 Torr, bath temp); IR (neat) 3386 (br), 2956, 1611, 1449, 1377, 1247, 996, 966, 866, 838, 727, 689 cm^{-1} ; NMR ($CDCl_3$) δ 0.08 (s, 9H), 0.99 (s, 6H), 1.73 (dm, $J = 6.0$ Hz, 3H), 3.75 (br.d, $J = 7.0$ Hz, 1H), 5.4-5.8 (m, 2H), 5.73 (d, $J = 19.2$ Hz, 1H), 6.02 (d, $J = 19.2$ Hz, 1H). Found: C, 67.99; H, 11.56%. Calcd for $C_{12}H_{24}OSi$: C, 67.86; H, 11.56%.

(E)-2,3,3-Trimethyl-5-(trimethylsilyl)-4-penten-2-ol: bp 60 °C (2 Torr, bath temp); IR (neat) 3448 (br), 2970, 1375, 1248, 839, 689 cm^{-1} ; NMR ($CDCl_3$) δ 0.05 (s, 9H), 1.03 (s, 6H), 1.15 (s, 6H), 5.70 (d, $J = 19.3$ Hz, 1H), 6.19 (d, $J = 19.3$ Hz, 1H). Found: C, 65.63; H, 12.35%. Calcd for $C_{11}H_{24}OSi$: C, 65.93; H, 12.07%.

A 1:1 Mixture of (E)-1-(Dimethylphenylsilyl)-1-pentene and 1-(Dimethylphenylsilyl)-2-pentene (stereoisomeric ratio 58:42): bp 67 °C (2 Torr, bath temp); IR (neat) 3004, 2958, 1427, 1248, 1113, 834, 726, 698 cm^{-1} ; NMR ($CDCl_3$) δ 0.26 (s,

1.74H), 0.27 (s, 1.26H), 0.32 (s, 3.0H), 0.89 (t, $J = 8.2$ Hz, 0.63H), 0.90 (t, $J = 7.0$ Hz, 0.58H), 0.93 (t, $J = 8.0$ Hz, 0.87H), 1.4-1.6 (m, 1H), 1.64 (dt, $J = 6.6, 1.1$ Hz, 1H), 1.71 (d, $J = 7.2$ Hz, 0.42H), 1.8-2.2 (m, 2H), 5.2-5.5 (m, 1.80H), 5.75 (dt, $J = 18.4, 1.4$ Hz, 0.50H), 6.12 (dt, $J = 18.4, 6.2$ Hz, 0.50H), 7.3-7.4 (m, 3H), 7.4-7.6 (m, 2H). Found: C, 76.54; H, 10.09%. Calcd for $C_{13}H_{20}Si$: C, 76.40; H, 9.86%.

(E)-2-Ethyl-4-(dimethylphenylsilyl)-1-phenyl-3-buten-1-ol (diastereomeric ratio 65:35): bp 123 °C (2 Torr, bath temp); IR (neat) 3396 (br), 3064, 2956, 2894, 1612, 1453, 1428, 1247, 1113, 1027, 995, 840, 826, 730, 698 cm^{-1} ; NMR ($CDCl_3$) δ 0.24 (s, 1.05H), 0.25 (s, 1.05H), 0.35 (s, 3.90H), 0.80 (t, $J = 7.6$ Hz, 1.95H), 0.87 (t, $J = 7.4$ Hz, 1.05H), 1.1-1.4 (m, 2H), 1.97 (d, $J = 4.4$ Hz, 0.35H), 2.09 (d, $J = 2.4$ Hz, 0.65H), 2.2-2.5 (m, 1H), 4.47 (dd, $J = 7.6, 2.4$ Hz, 0.65H), 4.61 (dd, $J = 6.4, 4.4$ Hz, 0.35H), 5.7-5.8 (m, 0.70H), 5.9-6.0 (m, 1.30H), 7.2-7.5 (m, 8H), 7.5-7.6 (m, 2H). Found: C, 77.52; H, 8.63%. Calcd for $C_{20}H_{26}OSi$: C, 77.36; H, 8.44%.

(E)-2-Methyl-1-phenyl-4-(dimethylphenylsilyl)-3-buten-4-one: bp 140 °C (2 Torr, bath temp); IR (neat) 3066, 2958, 1685, 1598, 1448, 1435, 1248, 1225, 1113, 991, 968, 845, 830, 730, 698 cm^{-1} ; NMR ($CDCl_3$) δ 0.28 (s, 3H), 0.30 (s, 3H), 1.34 (d, $J = 6.8$ Hz, 3H), 4.24 (ddq, $J = 7.2, 1.0, 6.8$ Hz, 1H), 5.97 (dd, $J = 18.7, 1.0$ Hz, 1H), 6.24 (dd, $J = 18.7, 7.2$ Hz, 1H), 7.2-7.7 (m, 8H), 8.0-8.1 (m, 2H). Found: C, 77.40; H, 7.57%. Calcd for $C_{19}H_{22}OSi$: C, 77.50; H, 7.53%.

(E)-2-Methyl-1-phenyl-4-(trimethylsilyl)-3-buten-1-one: bp 90 °C (2 Torr, bath temp); IR (neat) 2954, 1687, 1608, 1249, 1226, 990, 970, 867, 839, 699 cm^{-1} ; NMR ($CDCl_3$) δ 0.02 (s, 9H), 1.32 (d, $J = 6.8$ Hz, 3H), 4.20 (ddq, $J = 7.1, 1.1, 6.8$ Hz, 1H), 5.84 (dd, $J = 18.6, 1.1$ Hz, 1H), 6.14 (dd, $J = 18.6, 7.1$ Hz, 1H), 7.4-7.6 (m, 3H), 7.9-8.1 (m, 2H). Found: C, 72.22; H, 8.85%. Calcd for $C_{14}H_{20}OSi$: C, 72.36; H, 8.67%.

(E)-2,2-Dimethyl-1-phenyl-4-(trimethylsilyl)-3-buten-1-one: bp 95 °C (2 Torr, bath temp); IR (neat) 2956, 1682, 1599, 1249, 866, 839, 711, 696 cm^{-1} ; NMR ($CDCl_3$) δ 0.06 (s, 9H), 1.37 (s, 6H), 5.89 (d, $J = 19.1$ Hz, 1H), 6.33 (d, $J = 19.1$ Hz, 1H), 7.3-7.6 (m, 3H), 7.8-7.9 (m, 2H). Found: C, 73.27; H, 9.22%. Calcd for $C_{15}H_{22}OSi$: C, 73.11; H, 9.00%.

(E)-2-Ethyl-4-(dimethylphenylsilyl)-1-phenyl-3-buten-1-one: bp 117 °C (2 Torr, bath temp); IR (neat) 3064, 1958, 1872, 1682, 1598, 1448, 1427, 1249, 1217, 1113, 993, 836, 821, 730, 698 cm^{-1} ; NMR ($CDCl_3$) δ 0.29 (s, 3H), 0.31 (s, 3H), 0.93 (t, $J = 7.4$ Hz, 3H), 1.6-1.8 (m, 1H), 1.8-2.0 (m, 1H), 4.04 (dt, $J = 7.8, 7.4$ Hz, 1H), 5.98 (d, $J = 18.7$ Hz, 1H), 6.16 (dd, $J = 18.7, 7.8$ Hz, 1H), 7.2-7.6 (m, 8H), 7.9-8.1 (m, 2H). Found: C, 78.14; H, 8.06%. Calcd for $C_{20}H_{24}OSi$: C, 77.87; H, 7.84%.

Acknowledgement: Financial support by the Ministry of Education, Science, and Culture, (Grant-in-Aid for Special Project Research No. 59104005, 60219015, 61225011, and 62215017) is acknowledged.

References and Notes

1. Present address: Dept. Applied Chemistry, Okayama University of Science, Ridai-cho, Okayama 700, Japan.
2. Wiberg, E.; Stecher, O.; Andrascheck, H. J.; Kreuzbichler, L.; Stause, E. *Angew. Chem. Int. Ed.* 1963, **2**, 507; Rösch, L. *ibid.* 1977, **16**, 247, 480; Rösch, L.; Altnau, G.; Otto, W. H. *ibid.* 1981, **20**, 581; Rösch, L.; Starke, U. *ibid.* 1983, **22**, 557; Altnau, G.; Rösch, L. *Tetrahedron Lett.* 1983, **24**, 45; Trost, B. M.; Yoshida, J. *ibid.* 1983, **24**, 4895; Trost, B. M.; Yoshida, J.; Lautens, M. *J. Am. Chem. Soc.* 1983, **105**, 4494.
3. Altnau, G.; Rösch, L.; Bohlmann, F.; Lonitz, M. *Tetrahedron Lett.* 1980, **21**, 4069.
4. Fleming, I.; Newton, T. W.; Roessler, F. *J. Chem. Soc., Perkin Trans. 1* 1981, 2527.
5. Tamao, K.; Akita, M.; Kanatani, R.; Ishida, N.; Kumada, M. *J. Organomet. Chem.* 1982, **226**, C9.
6. Hayami, H.; Sato, M.; Kanemoto, S.; Morizawa, Y.; Oshima, K.; Nozaki, H. *J. Am. Chem. Soc.* 1983, **105**, 4491.
7. Nozaki, K.; Wakamatsu, K.; Nonaka, T.; Tückmantel, W.; Oshima, K.; Utimoto, K. *Tetrahedron Lett.* 1986, **27**, 2007.
8. Transition-metal catalyzed hydrosilylation of acetylenes: Negishi, E. "Organometallics in Organic Synthesis," Vol. 1, John Wiley and Sons, New York, 1980, p. 406; Birkofer, L.; Kühn, T.; *Chem. Ber.* 1978, **111**, 3119; Tamao, K.; Yoshida, J.; Yamamoto, H.; Kakui, T.; Matsumoto, H.; Takahashi, M.; Kurita, A.; Murata, M.; Kumada, M. *Organometallics*, 1982, **1**, 355.
9. Tückmantel, W.; Oshima, K.; Nozaki, H. *Chem. Ber.* 1986, **119**, 1581; Wakamatsu, K.; Nonaka, T.; Okuda, Y.; Tückmantel, W.; Oshima, K.; Utimoto, K.; Nozaki, H. *Tetrahedron* 1986, **42**, 4427.
10. Posner, G. H. *Org. React.* 1972, **19**, 1; Faust, J.; Froböse, R. *Gmelin Handbook of Inorganic Chemistry*, 8th ed. "Organocopper Compounds Part 2", Springer, Berlin, 1983; Normant, J. F.; Alexakis, A. *Synthesis* 1981, 841.
11. Normant, J. F. Organocopper and Organomanganous reagent in "Modern Synthetic Methods" Vol. 3, Scheffold, R. Ed. Otto Salle Verlag GmbH Co. Frankfurt am Main, 1983, p.139; Kauffmann, T.; Bisling, M. *Tetrahedron Lett.* 1984, **25**, 293.
12. Watanabe, H.; Kobayashi, M.; Sato, M.; Nagai, Y. *J. Organomet. Chem.* 1981, **216**, 149; Watanabe, H.; Kobayashi, M.; Higuchi, K.; Nagai, Y. *ibid.* 1980, **186**, 51; Matsumoto, H.; Matsubara, I.; Kato, T.; Shono, H.; Watanabe, H.; Nagai, Y. *ibid.* 1980, **199**, 43; Sakurai, H.; Kamiyama, Y.; Nakadaira, Y. *J. Am. Chem. Soc.* 1975, **97**, 931; Okinoshima, H.; Yamamoto, K.; Kumada, M. *ibid.* 1972, **94**, 9263; Kusumoto, T.; Hiyama, T. *Tetrahedron Lett.* 1987, **28**, 1807.
13. Hibino, J.; Nakatsukasa, S.; Fugami, K.; Matsubara, S.; Oshima, K.; Nozaki, H. *J. Am. Chem. Soc.* 1985, **107**, 6416.
14. Disilylation of acetylenes can sometimes be achieved using metal/silyl halide systems. Dunogues, J.; Bourgeois, P.; Pillot, J.-P.; Merault, G.; Calas, R. *J. Organomet. Chem.* 1975, **87**, 169 and references cited therein.
15. Still, W. C. *J. Org. Chem.* 1976, **41**, 3063.
16. We are tempted to assume that the active reagent in our method could be $(\text{Me}_3\text{Si})_3\text{MnMgMe}$ derived from 3 equiv of Me_3SiMgMe and MnCl_2 . The addition of PhCHO to the reagent provided only phenyl(trimethylsilyl)carbinol and no

trace of 1-phenylethanol.

17. Stannylmetalation of acetylenes has been reported. Hibino, J.; Matsubara, S.; Morizawa, Y.; Oshima, K.; Nozaki, H. Tetrahedron Lett. **1984**, 25, 2151.
18. In the case of terminal acetylenes and silylacetylenes in Table 1, the intermediary alkenylmanganese could not be trapped by electrophiles such as D₂O and MeI.
19. Oda, H.; Sato, M.; Morizawa, Y.; Oshima, K.; Nozaki, H. Tetrahedron **1985**, 41, 3257.
20. Sakurai, H.; Nakadaira, Y.; Kira, M.; Tobita, H. Tetrahedron Lett. **1980**, 21, 3077; Chung, C.; Lagow, R. J. J. Chem. Soc., Chem. Commun. **1972**, 1078; Sakurai, H.; Nakadaira, Y.; Tobita, H. J. Am. Chem. Soc. **1982**, 104, 300; idem, Chem. Lett. **1982**, 771.
21. For reviews see: Chan, T. H.; Fleming, I. Synthesis **1979**, 761; Sakurai, H. Pure Appl. Chem. **1982**, 54, 1; Fleming, I. Comp. Org. Chem. **1979**, 3, 539; Magnus, P. D.; Sarkar, T.; Djuric, S. Comp. Organomet. Chem. **1982**, 7, 515; Magnus, P. D. Aldrichimica Acta. **1980**, 13, 43.
22. Okuda, Y.; Sato, M.; Oshima, K.; Nozaki, H. Tetrahedron Lett. **1983**, 24, 2015.
23. Fugami, K.; Oshima, K.; Utimoto, K.; Nozaki, H. ibid. **1986**, 27, 2161.
24. Corey, E. J.; Posner, G. H. ibid. **1970**, 315.
25. Transformation of alkenyl halides into vinylsilanes by Wurtz-type coupling has been reported. For example, see Chan, T. H.; Baldassarre, A.; Massuda, D. Synthesis **1976**, 801; Nagendrappa, C. ibid. **1980**, 704; Fleming, I.; Pearce, A. J. Chem. Soc., Perkin 1, **1980**, 2485. Metal-halogen exchange followed by electrophilic attack by R₃SiCl has provided another route to vinylsilanes. Neumann, H.; Seebach, D. Tetrahedron Lett. **1976**, 4839.
26. Conversion of allyl ethers and thio ethers to allylsilanes using metal/silyl halide systems has been reported. Leading references are: Deleris, G.; Kowalski, J.; Dunegues, J.; Calas, R. Tetrahedron Lett. **1977**, 4211; Pandyszekeres, D.; Dèlèris, G.; Picard, J-P.; Pillot, J-P.; Calas, R. ibid. **1980**, 21, 4267; Biran, C.; Dunogues, J.; Calas, R.; Gerval, J.; Tskhovrebachvili, T. Synthesis, **1981**, 220; Tzeng, D.; Weber, W. P. J. Org. Chem. **1981**, 46, 265.
27. Cross coupling reactions between allylic sulfides and Grignard reagents in the presence of transition-metal catalyst have been reported. Gendreau, Y.; Normant, J. F.; Villieras, J. J. Organomet. Chem. **1977**, 142, 1; Barsanti, P.; Calo, V.; Lopez, L.; Marchese, G.; Naso, F.; Peace, G. J. Chem. Soc., Chem. Commun. **1978**, 1085; Okamura, H.; Takei, H. Tetrahedron Lett. **1979**, 3425.
28. Fugami, K.; Nakatsukasa, S.; Oshima, K.; Utimoto, K.; Nozaki, H. Chem. Lett. **1986**, 869.
29. Silyltitanation of isoprene has been reported to give 4-dimethylphenylsilyl-2-methyl-1-butene. See ref. 5.
30. A stock solution in THF was used. Gilman, H.; Lichtenwalter, G. D. J. Am. Chem. Soc. **1958**, 80, 607.
31. Anhydrous manganese(II) chloride and a stock solution of Li₂MnCl₄ in THF are equally effective for the reaction.
32. Fuchigami, T.; Ojima, I. J. Organomet. Chem. **1981**, 212, 145.